

Raman and NEXAFS study of the oxidation of Cu₃N thin films during their growth procedure

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Cu₃N thin films were grown by annealing of Cu under NH₃:O₂ flow. The Cu layers were deposited on different substrates by sputtering. The procedure followed is the one proposed by Matsuzaki *et al.* [1], where the O₂ molecules adsorbed on the Cu surface are transformed to oxygen species that selectively cause dehydrogenation of NH₃ creating adatoms on the surface that migrate into the bulk forming Cu₃N. The annealing temperatures were ranging from 300 to 800 °C. The samples were studied by Raman and N-K-edge Near Edge X-ray Absorption Fine Structure (NEXAFS) spectroscopies. Raman spectra were recorded using two different excitation wavelengths (647.1 and 514.5 nm) since resonance effects affect, to a different extent, the Raman peak intensities of copper oxides and Cu₃N, allowing for their better discrimination. Cu₂O surface oxide is formed in almost all the samples even at low annealing temperatures. On the other hand, the formation of CuO is generally observed at higher temperatures with the onset depending on the amount of O₂ in the gas mixture. However, high temperature is necessary to grow larger crystallites, as it is deduced by the narrowing of the high-frequency Raman peak (at ca. 650 cm⁻¹) of Cu₃N and SEM. N-K-edge NEXAFS spectra revealed that the formation of CuO, at elevated annealing temperatures, is accompanied by the creation of N₂ trapped in the sample. Molecular nitrogen is most probably formed by the N atoms originating from the dissociation of the Cu-N bonds.

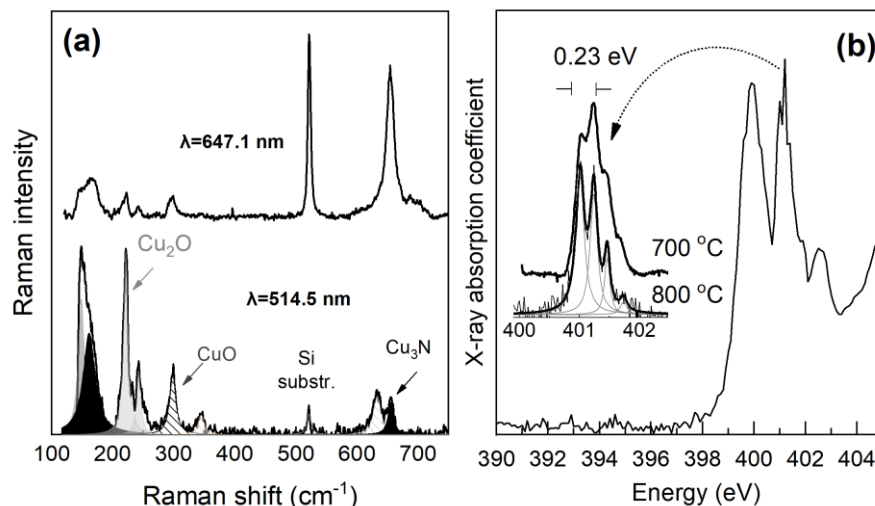


Figure 1: (a) Raman spectra of a sample annealed at 600 °C recorded using two different excitation wavelengths. (b) N-K-edge NEXAFS spectrum of a sample annealed at 700 °C. The inset shows the fine structure of the N₂ peak due to vibronic transitions.

Reference

[1] K. Matsuzaki, K. Harada, Y. Kumagai, S. Koshiya, K. Kimoto, S. Ueda, M. Sasase, A. Maeda, T. Susaki, M. Kitano, F. Oba, H. Hosono, *Adv. Mater.* **30**, 1801968 (2018).

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